

# Total Synthesis and Complete Stereostructure of Gambieric Acid A

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**S** Supporting Information

**ABSTRACT:** Total synthesis of gambieric acid A, a potent antifungal polycyclic ether metabolite, has been accomplished for the first time, which firmly established the complete stereostructure of this natural product.

Marine polycyclic ether natural products constitute one of the most intriguing families of secondary metabolites known to date.<sup>1,2</sup> This intrigue is due primarily to their highly complex molecular architecture, potent biological activities, and extreme natural scarcity. Gambieric acids (GAs) A–D (Figure 1) were isolated from the cultured cells of the ciguatera

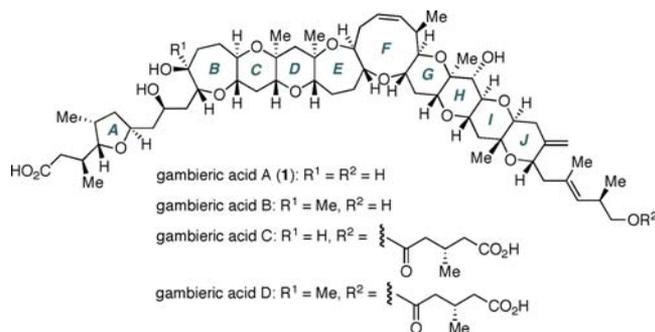


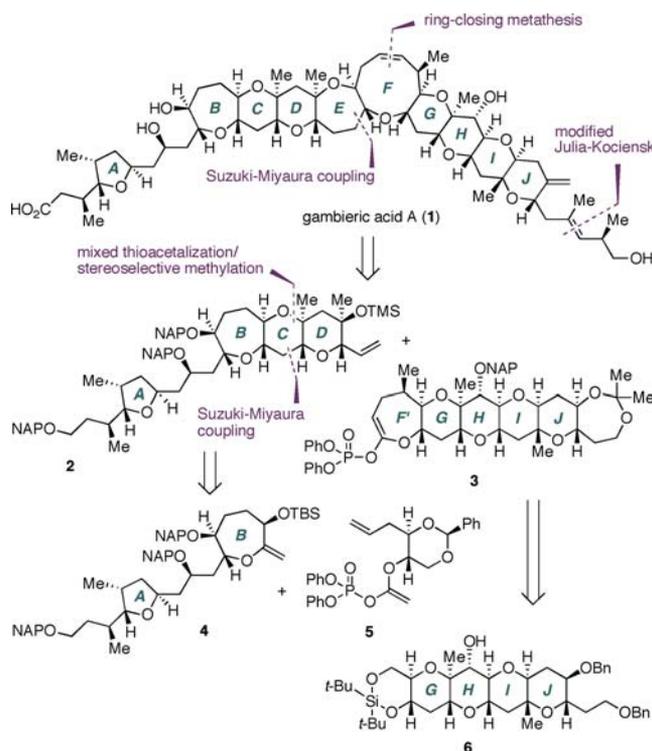
Figure 1. Structures of gambieric acids A–D.

causative dinoflagellate *Gambierdiscus toxicus* (GIII strain) by Nagai, Yasumoto, and co-workers.<sup>3</sup> The gross structure, including the relative stereochemistry of the polycyclic ether domain, was determined through extensive 2D NMR spectroscopic analyses. The complete stereostructure of GAs was assigned on the basis of NMR-based conformational analysis, degradation experiments, the application of chiral anisotropic reagents, and chiral HPLC analysis.<sup>4</sup> However, our own studies on model compounds have suggested that the absolute configuration of the polycyclic ether domain of GAs is opposite to that of the originally assigned structure.<sup>5</sup> Accordingly, we proposed the stereostructure of GAs shown in Figure 1; however, this structure needed to be confirmed by total synthesis. (+)-Gambieric acid A (**1**) exhibits exceedingly potent antifungal activity. Specifically, **1** is approximately 2000 times more potent than amphotericin B. It is known that **1** only weakly displaces tritiated dihydrobrevetoxin-B (<sup>3</sup>H]PbTx-3) from voltage-gated sodium ion channels<sup>6</sup> and does not show any detectable toxicity against mice at 1 mg/kg, which was the highest dose tested,<sup>7</sup> even though **1** is structurally similar to ion channel-activating polycyclic ether neurotoxins, such as brevetoxins and ciguatoxins.<sup>1</sup> Meanwhile, Nagai et al. suggested

that **1** has a possible role as an endogenous growth-regulating factor of *G. toxicus*.<sup>8</sup> However, detailed investigations into the biological mode-of-actions of GAs have been precluded by their limited availability from natural sources. Although recent methodological advances<sup>2</sup> have greatly improved our ability to synthesize complex polycyclic ether molecules, GAs remain unconquered because of their anomalous structural complexity, which poses a formidable synthetic challenge to organic chemists.<sup>9,10</sup> Here, we disclose the first total synthesis of **1**, which resulted in unambiguous establishment of its complete stereostructure.

Our synthetic plan toward **1** is illustrated in Scheme 1. We planned to introduce the J-ring side chain at a late stage of the

## Scheme 1. Synthesis Plan toward **1**



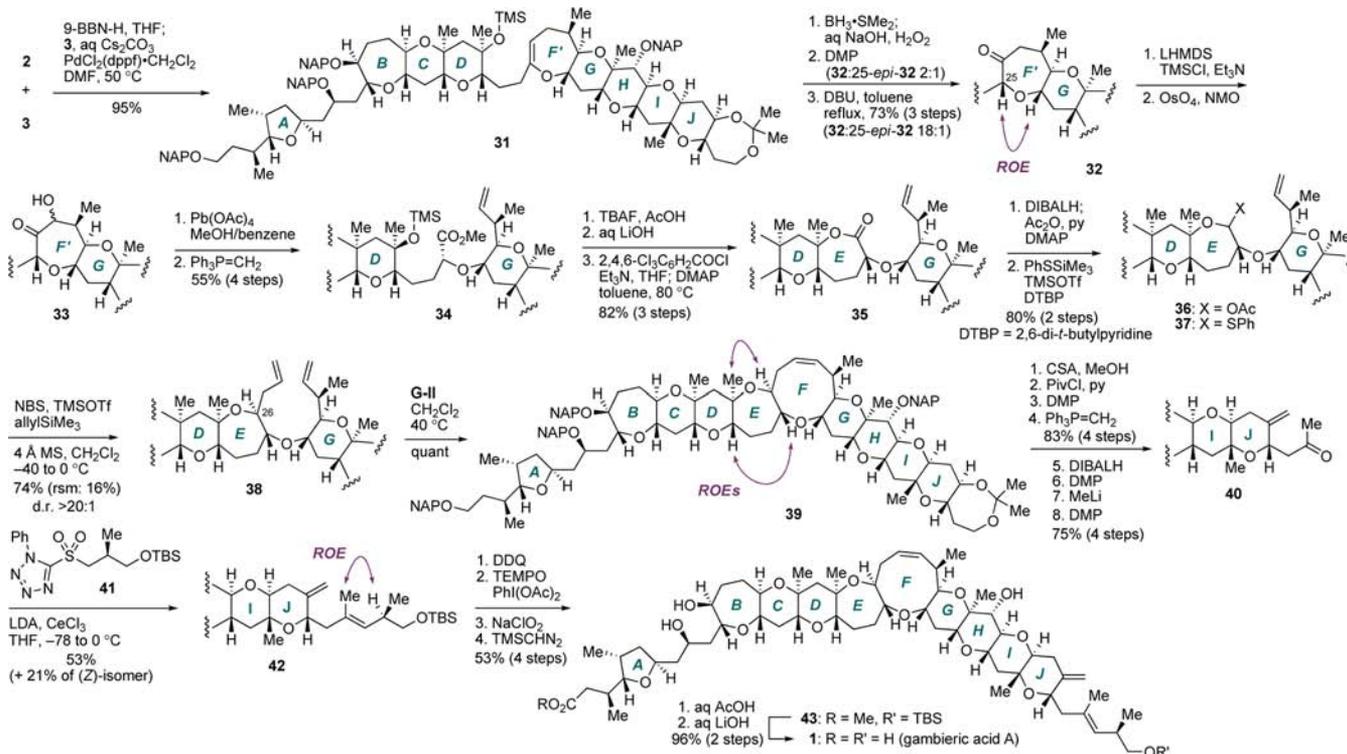
total synthesis, by using the modified Julia–Kocienski olefination.<sup>10g,h,11</sup> We retrosynthetically divided the nonacyclic core structure of **1** into the A/BCD-ring fragment **2** and F'GHIJ-ring fragment **3** containing a contracted F'-ring moiety. These two intermediates of equal structural complexity would

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Scheme 4. Completion of the Total Synthesis of 1



on the basis of NOE experiments as shown. To complete the A/BCD-ring fragment, the C21 axial methyl group had to be introduced onto the D-ring. To this end, 27 was elaborated to *exo*-olefin 28 in four steps. Dihydroxylation of 28 proceeded from the sterically less encumbered face of the molecule to give 1,2-diol 29 as a single stereoisomer.<sup>16a</sup> Tosylation of 29 and subsequent LiEt<sub>3</sub>BH reduction uneventfully afforded tertiary alcohol 30. Finally, desilylation, oxidation, methylenation, and silylation furnished the A/BCD-ring fragment 2.

The completion of the total synthesis of 1 is illustrated in Scheme 4. First, we assembled the A/BCD- and F'GHIJ-ring fragments (2 and 3,<sup>22</sup> respectively) through a Suzuki–Miyaura coupling, which afforded endocyclic enol ether 31 in a respectable yield. Hydroboration of 31 delivered a 2:1 mixture of diastereomeric alcohols. Without separation, these alcohols were oxidized to give an epimeric mixture of ketones 32 and 25-*epi*-32 (not shown), which was then treated with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) to promote epimerization at the C25 stereogenic center. This allowed us to convert the most of the unwanted 25-*epi*-32 into the desired 32, the latter being thermodynamically favored under equilibrating conditions (32:25-*epi*-32 = 18:1, after epimerization). The C25 stereochemistry was confirmed by an ROE experiment, as shown. Having successfully established the C25 stereogenic center, the F'-ring was oxidatively cleaved via the intermediacy of  $\alpha$ -hydroxy ketone 33. Thus, dihydroxylation of the silyl enol ether derived from 32 gave 33, which was exposed to Pb(OAc)<sub>4</sub> in methanol/benzene, and then immediately methylenated to deliver ester 34. Desilylation, followed by saponification of the methyl ester, and ensuing lactonization<sup>24</sup> afforded lactone 35. Reductive acetylation<sup>25</sup> of 35 gave  $\alpha$ -acetoxy ether 36. Our previous studies on model compounds suggested that harsh reaction conditions would be necessary for direct allylation of 36 due to its low reactivity.<sup>10a–c,e,f</sup> Instead,

we envisaged an application of our glycosylation chemistry<sup>26</sup> to the corresponding mixed thioacetal 37. Thus, stereoselective allylation to complete the E-ring was best accomplished by exposing 37 to allyltrimethylsilane in the presence of NBS/trimethylsilyl trifluoromethanesulfonate (TMSOTf) to provide diene 38 in 74% yield as a single stereoisomer with correct C26 stereochemistry. RCM of 38 using G-II cleanly furnished nonacyclic core 39 quantitatively. The relative configurations of the stereogenic centers around the E- and F-rings were confirmed at this stage by the ROE experiment as shown.

The J-ring side chain needed to be constructed to complete the total synthesis. Toward this end, 39 was converted to methyl ketone 40 via an eight-step sequence. Julia–Kocienski olefination of 40 using sulfone 41<sup>10g,h</sup> was performed in the presence of CeCl<sub>3</sub><sup>10g,h</sup> to afford the desired trisubstituted olefin 42 together with the corresponding (*Z*)-isomer (not shown). These isomers were separated by flash chromatography using silica gel. Oxidative cleavage of the 2-naphthylmethyl (NAP) ethers, selective two-stage oxidation of the liberated primary hydroxy group, and subsequent esterification provided methyl ester 43. Finally, acidic cleavage of the silyl ether and saponification of the methyl ester furnished (+)-gambieric acid A (1). The spectroscopic properties (IR, <sup>1</sup>H, <sup>13</sup>C NMR, and HRMS) and optical rotation value of synthetic (+)-1 ([ $\alpha$ ]<sub>D</sub><sup>25</sup> +22.5 (c 0.40, MeOH)) were in full accordance with those of the authentic sample ([ $\alpha$ ]<sub>D</sub><sup>20</sup> +33 (c 0.488, MeOH)).<sup>3</sup> Furthermore, the synthetic material displayed antifungal activity against *Aspergillus niger*, which was equipotent to that of the natural product. Thus, we conclude that the structure 1, shown in Figure 1, represents the complete stereostructure of (+)-gambieric acid A, confirming our previous stereochemical reassignment on the basis of the synthesis and spectroscopic analysis of model compounds.<sup>5</sup>

In conclusion, we have completed the total synthesis of (+)-gambieric acid A (**1**) for the first time by exploiting synthetic methodologies developed within our laboratory. The chemistry described herein opens up avenue for the preparation of synthetic analogues that would be helpful for addressing the biological profile of **1**, which is uniquely different from that of ion channel-activating polycyclic ether neurotoxins despite the high structural similarity shared among this class of natural products. This study also demonstrated the vital role of total synthesis in the structure determination of complex natural products.

## ■ ASSOCIATED CONTENT

### Supporting Information

Experimental procedures and spectroscopic data for all new compounds and copies of  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra for selected compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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### Notes

The authors declare no competing financial interests.

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